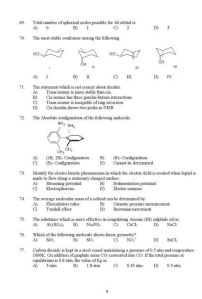




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Q3: Use the following information to answer the next question.

An anion has *i* electrons in its shell than the number of protons in its nuclei, whereas a cation has *ii* protons in its nuclei than the number of electrons in its shell.

The information in which alternative completes the given statement?

- A. *i* / *less* / *less*
- B. *i* / *less* / *more*
- C. *i* / *more* / *more*
- D. *i* / *more* / *less*

Types of Chemical Reactions Worksheet #1

H/Chemistry

For each of the following equations, identify what kind of reaction it represents: double replacement, single replacement, decomposition, or synthesis (composition).

- 1) $2Mg + O_2 \rightarrow 2MgO$ **Synthesis**
- 2) $CaI_2 + Cl_2 \rightarrow CaCl_2 + I_2$ **Single Replacement**
- 3) $3KOH + AlCl_3 \rightarrow Al(OH)_3 + 3KCl$ **Double Replacement**
- 4) $C + O_2 \rightarrow CO_2$ **Synthesis**
- *5) $Ca(OH)_2 + 2HCl \rightarrow CaCl_2 + 2H_2O$ **Double Replacement**
- 6) $2KClO_3 \rightarrow 2KCl + 3O_2$ **Decomposition**
- 7) $3CuSO_4 + 2Al \rightarrow Al_2(SO_4)_3 + 3Cu$ **Single Replacement**
- 8) $Na_2S + 2AgNO_3 \rightarrow 2NaNO_3 + Ag_2S$ **Double Replacement**
- 9) $3H_2 + N_2 \rightarrow 2NH_3$ **Synthesis**
- *10) $2Na + 2H_2O \rightarrow 2NaOH + H_2$ **Single Replacement**

*Remember: "H₂O" can also be thought of as "HOH"

Complete each of the following equations as needed to make it the type of reaction indicated. Be sure to write each formula correctly.

- 11) Double replacement: $Na_2CrO_4 + PbCl_2 \rightarrow 2NaCl + PbCrO_4$
- 12) Single replacement: $Cl_2 + 2NaBr \rightarrow 2NaCl + Br_2$
- 13) Decomposition: $Mg(ClO_3)_2 \rightarrow MgCl_2 + 3O_2$
- 14) Synthesis: $2H_2 + O_2 \rightarrow 2H_2O$
- 15) Double replacement: $3Ca(OH)_2 + 2FeCl_3 \rightarrow 2Fe(OH)_3 + 3CaCl_2$
- 16) Single replacement: $Fe + Cu(NO_3)_2 \rightarrow Fe(NO_3)_2 + Cu$ [Assume Fe²⁺]
- 17) Decomposition: $2Hg_2O \rightarrow 4Hg + O_2$
- 18) Synthesis: $S + O_2 \rightarrow SO_2$
- 19) Double replacement: $AgNO_3 + KI \rightarrow AgI + KNO_3$
- 20) Single replacement: $Cu + 2AgNO_3 \rightarrow 2Ag + Cu(NO_3)_2$ [Copper (II) is used here]

A diagram of the reactions of the first round of fatty acid synthesis (lipogenesis) is shown below.

How many cycles of the synthesis pathway are needed to produce palmitic acid, C₁₆H₃₂COOH?

Number

CH₃-C(=O)-S-KS + HO-C(=O)-CH₂-C(=O)-S-ACP

CO₂ ↓

CH₃-C(=O)-CH₂-C(=O)-S-ACP

NADPH + H⁺ ↓

CH₃-CH(OH)-CH₂-C(=O)-S-ACP

H₂O ↓

CH₃-CH=CH-C(=O)-S-ACP

NADPH + H⁺ ↓

NADP⁺ ↓

Part B

Demonstrate the definition with a chemical equation for HCl

Express your answer as a chemical equation. Identify all of the phases in your answer.

Submit Previous Answers Request Answer

Incorrect; Try Again; 7 attempts remaining

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